

## PALLADIUM CATALYZED MONO- AND DIARYLATION OF 2-METHYLQUINOLINES

**Edgars Abele**, Ramona Ābele, Ļena Golomba

*Latvian Institute of Organic Synthesis, 21 Aizkraukles Street, Riga, LV-1006, Latvia,  
E-mail: abele@osi.lv*

### ABSTRACT:

Reactions of 2-methylquinolines with aryl iodides in the system Pd(OAc)<sub>2</sub> (5 mol.%)/ dppb (1,4-bis(diphenylphosphino)butane) (10 mol.%) / *t*-BuOK (1.3 eq.) in toluene afforded mono- and diarylated products in yields up to 44 %.

**Keywords:** Palladium catalyst, 2-methylquinolines, 2-benzyl-6-methoxy-quinoline, 6-Methoxy-2-(diphenylmethyl)quinoline, 6-methoxy-2-(4-methoxybenzyl)quinoline

### INTRODUCTION

Benzylquinolines and diarylmethylquinolines are of interest as valuable intermediates in organic synthesis. Usually benzylquinolines were obtained by reaction of benzyl sodium with quinoline<sup>1</sup>, benzyl magnesium bromide with quinoline<sup>2</sup>, methylquinoline with chlorobenzene<sup>3</sup> or by homolytic benzylation of quinoline with toluene derivatives<sup>4</sup>.

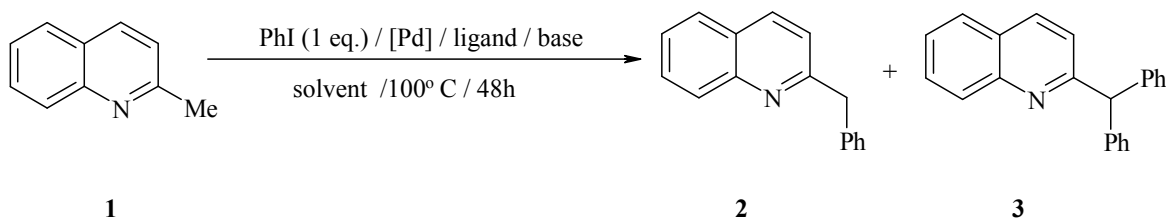
Diarylmethylquinolines were obtained from quinoline and benzene in the presence of dehydrating agent (for example, H<sub>2</sub>SO<sub>4</sub>)<sup>5</sup> or from quinoline aldehyde and benzene in the presence of triflic acid<sup>6</sup>.

Palladium catalyzed arylation reactions of ketones, amides and related nucleophiles were recently described in some reviews<sup>7-9</sup>. The best catalytic systems for arylation of ketone enolates were found to be Pd<sub>2</sub>(dba)<sub>3</sub> / BINAP / *t*-BuONa<sup>10</sup>, Pd<sub>2</sub>(dba)<sub>3</sub> / Tol-BINAP / *t*-BuONa / THF<sup>11</sup>, Pd(OAc)<sub>2</sub> / Xantphos / *t*-BuONa or K<sub>3</sub>PO<sub>4</sub><sup>12</sup>, Pd(OAc)<sub>2</sub> / PPh<sub>3</sub> / Cs<sub>2</sub>CO<sub>3</sub> / DMF<sup>13</sup>, Pd(OAc)<sub>2</sub> / RR<sub>3</sub> (R= alkyl, aryl) / Cs<sub>2</sub>CO<sub>3</sub><sup>14</sup> and (imidazol-2-ylidene)palladium acetate / *t*-BuONa / dioxane<sup>15</sup>. Palladium catalyzed arylation of nitriles<sup>16</sup> and sulfones<sup>17</sup> were described too. Recently was described simple arylation of 8-methylquinoline with 4-bromo-1-iodobenzene to corresponding benzylquinoline in the system Pd(OAc)<sub>2</sub> / AgOAc / AcOH<sup>18</sup>. However, only mono-arylated products were obtained in the above system. Recently we reported our first results in Pd-catalyzed mono- and diarylation of methylquinolines.<sup>19</sup> Now we are presenting a novel Pd-catalyzed method of synthesis of benzylquinolines and diarylmethylquinolines.

### RESULTS AND DISCUSSION

We have developed a new and simple palladium catalyzed arylation method of methylquinolines. The influence of catalyst, base and solvent was studied in the arylation

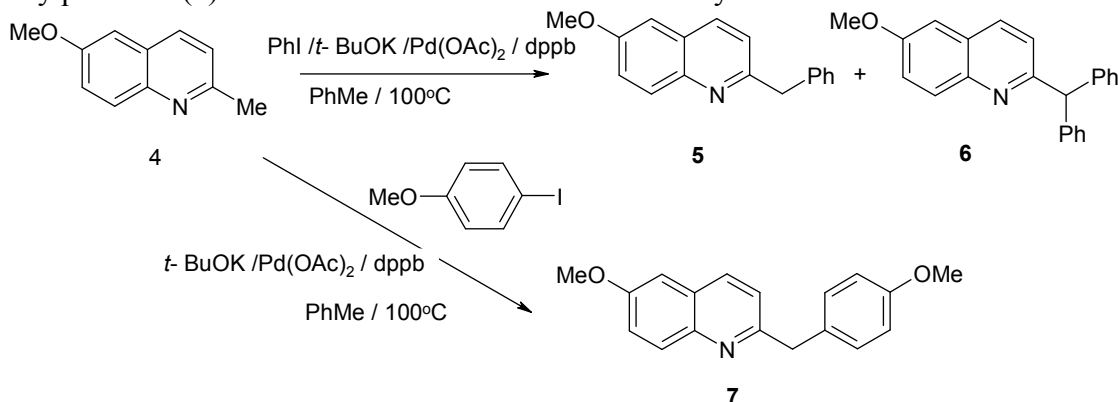
reaction of 2-methylquinoline (**1**) with iodobenzene (2 eq.). Interestingly, the system Pd(OAc)<sub>2</sub> (5 mol.%)/ dppb (1,4-bis(diphenylphosphino)butane) (10 mol.%) / *t*-BuOK (1.3 eq.) in toluene was found to be the most active for the diarylation of compound **1**. In this system compound **3** was obtained in the 41% yield, along with mono-arylated product **2** (17%). Increase in the amount of *t*-BuOK to 2.6 equiv. and using P(*o*-Tol)<sub>3</sub> as ligand diminishes the yields of products **2** and **3**. The systems Pd(OAc)<sub>2</sub> (5 mol.%) / CuI (10 mol.%) / *t*-BuOK / toluene, Pd(OAc)<sub>2</sub> (5 mol.%) / dppb (10 mol.%) / *t*-BuOK (1.3 eq.) / CsOH (1.3 eq.) / 18-crown-6 (10 mol.%) / toluene and Pd(PPh<sub>3</sub>)<sub>4</sub> (5mol.%) / *t*-BuOK (1.3 eq.) / toluene were essentially inactive in the arylation of compound **1** (Table 1).



**Table 1. Palladium catalyzed arylation of 2-methylquinoline (**1**) with 2 equivalent of PhI at 100°C for 48 h.**

Catalyst	Ligand	Base	Solvent	Yield of <b>2</b> , % (GC-MS data)	Yield of <b>3</b> , % (GC-MS data)
Pd(OAc) <sub>2</sub> (5 mol.%)	Dppb (10 mol.%)	<i>t</i> -BuOK (1.3 eq.)	Toluene	17	41
Pd(OAc) <sub>2</sub> (2.5 mol.%)	Dppb (5 mol.%)	<i>t</i> -BuOK (1.3 eq.)	Toluene	26	0
Pd(OAc) <sub>2</sub> (5 mol.%)	Dppb (10mol.%)	<i>t</i> -BuOK (2.6 eq.)	Toluene	13	5
Pd(OAc) <sub>2</sub> (5 mol.%)	Dppb (10 mol.%)	<i>t</i> -BuOK (1.3 eq.)	DMF	23	0
Pd(OAc) <sub>2</sub> (5 mol.%)	Dppb (10 mol.%)	CsOH (1.3 eq.) + 18-crown-6 (10 mol.%)	Toluene	0	0
Pd(OAc) <sub>2</sub> (5 mol.%)	P( <i>o</i> -Tol) <sub>3</sub> (10mol.%)	<i>t</i> -BuOK (1.3 eq.)	Toluene	17	7
Pd(OAc) <sub>2</sub> (5 mol.%) + CuI (10 mol.%)	Dppb (10mol.%)	<i>t</i> -BuOK (1.3 eq.)	Toluene	2	0
Pd(PPh <sub>3</sub> ) <sub>4</sub> (5 mol.%)	-	<i>t</i> -BuOK (1.3 eq.)	Toluene	Traces	0

The catalytic system system Pd(OAc)<sub>2</sub> (5 mol.%)/ dppb (1,4-bis(diphenylphosphino)butane) (10 mol.%) / *t*-BuOK (1.3 eq.) in toluene, as the most active, was used in the arylation of 2-methyl-6-methoxyquinoline (**4**) with iodobenzene or 1-iodo-4-methoxybenzene.



Arylation of quinoline **4**, having electron donating group in the position 6, with iodobenzene in above system leads to mixture of mono- (**5**) (yield 12%) and diarylated product (**6**) (yield 13%). However, reaction of compound **4** with 1-iodo-4-methoxybenzene was selective so that 6-methoxy-2-(4-methoxybenzyl)quinoline (**7**) was isolated in 44 % yield as single product. Diarylation of compound **4** in this case did not occurred due to deactivation of methylene group in the compound **7** by methoxy group in *para*-position.

Thus, palladium catalyzed arylation of methylquinolines is a simple method for the synthesis of benzylquinolines and diarylmethylquinolines which otherwise are difficult to obtain.

## EXPERIMENTAL SECTION

<sup>1</sup>H spectra were recorded on a Varian Mercury BB 400 MHz in CDCl<sub>3</sub> using HMDS as internal standard. Mass spectra were registered on a GC-MS HP 6890 (70 eV).

### General procedure for the arylation of 2-methylquinoline with iodobenzene in the presence of palladium catalyst

2-Methylquinoline (0.14 ml, 1 mmol), iodobenzene (0.22 ml, 2 mmol) and base (see Table 1) were added to stirred solution of palladium catalyst and ligand (see Table 1) in dry toluene (1.25 ml) in a Pierce reacti-vial (5 ml) under argon atmosphere. The mixture was stirred at 100°C (GC-MS control) for 48 h, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: toluene: ethyl acetate 10:1) to obtain desired products **2** and **3**<sup>5,6</sup> as colorless oils (see Table 1).

### General procedure for the arylation of 2-methyl-6-methoxyquinoline (**4**) with iodobenzene or 1-iodo-4-methoxybenzene in the presence of palladium catalyst.

2-Methyl-6-methoxyquinoline (**4**) (0.224 g, 1 mmol), aryl iodide (2 mmol) and *t*-BuOK (0.146 g, 1.3 mmol) were added to stirred solution of Pd(OAc)<sub>2</sub> (0.011 g, 0.05 mmol) and 1,4-bis(diphenylphosphino)butane (dppb) (0.043 g, 0.1 mmol) in dry toluene (1.25 ml) in a Pierce reacti-vial (5 ml) under argon atmosphere. The mixture was stirred at 100°C (GC-MS control) for 48 h, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: toluene: ethyl acetate 10:1) to obtain desired products **5-7** as colourless oils.

The properties of obtained products were as follows:

**2-Benzyl-6-methoxy-quinoline (5).** Yield 12 %. <sup>1</sup>H NMR δ (ppm): 3.91 (s, 3H, Me), 4.30 (s, 2H, CH<sub>2</sub>), 7.02-7.38 and 7.90-8.00 (both m, 10H, Ph and quinoline ring protons). MS: *m/z* (%): 249 (M<sup>+</sup>, 58), 248 (100), 234 (10), 205 (21).

**6-Methoxy-2-(diphenylmethyl)quinoline (6).** Yield 13 %. <sup>1</sup>H NMR δ (ppm): 3.91 (s, 3H, Me), 5.87 (s, 1H, CH), 7.03-7.31 and 7.93-7.97 (both m, 15H, Ph and quinoline ring protons). MS: *m/z* (%): 326 (M<sup>+</sup>, 15), 324 (100), 281 (16), 204 (26), 165 (33), 152 (13).

**6-Methoxy-2-(4-methoxybenzyl)quinoline (7).** Yield 44 %. <sup>1</sup>H NMR δ (ppm): 3.76 (s, 3H, MeO in Ph ring) 3.91 (s, 3H, MeO in quinoline ring), 4.23 (s, 2H, CH<sub>2</sub>), 6.81-7.37 and 7.89-7.99 (both m, 9H, Ph and quinoline ring protons). MS: *m/z* (%): 279 (M<sup>+</sup>, 98), 278 (100), 264 (66), 221 (11), 192 (12), 121 (14).

## ACKNOWLEDGEMENTS

This work was supported by the project of ESF Foundation of Latvia (Project Nr. 2009/0197/1DP/1.1.1.2.0/09/APIA/VIAA/014).

## REFERENCES

1. H. Gilman, J.A. Beel, J. Amer. Chem. Soc. 73, 774 (1951).
2. E. Bergman, J. Rosenthal, J. Prakt. Chem. 135, 267 (1932).
3. R.E. Wrigh, F.W. Bergstrom, J. Org. Chem. 1, 179 (1936).
4. A. Clerici, O. Porta, Canad. J. Chem. 58, 2117 (1980).
5. W. Mathes, A. Wolf, US Pat. 3042678 (1962); Chem. Abstr. 57, 15082e (1962).
6. D.A. Klumpp, A. Jones, S. Lau, S. de Leon, Synthesis 1117 (2000).
7. D. Prim, J.-M. Campagne, D. Joseph, B. Andrioletti, Tetrahedron 58, 2041 (2002).
8. C. Scolastino, P. Giovanni, Chemtracts 12, 498 (1999).
9. C. Bolm, J. P. Hildebrand, K. Muniz, N. Hermanns, Angew. Chem. Int. Ed. 40, 3284 (2001).
10. J. Ahman, J.P. Wolfe, M.V. Troutman, M. Palucki, S.L. Buchwald, J. Amer. Chem. Soc. 120, 1918 (1998).
11. M. Palucki, S.L. Buchwald, J. Amer. Chem. Soc. 119, 11108 (1997).
12. J. M. Fox, X. Huang, A. Chieffi, S. L. Buchwald, J. Amer. Chem. Soc. 122, 1360 (2000).
13. F. Churruca, R. SanMartin, M. Carril, I. Tellitu, E. Dominguez, Tetrahedron 60, 2393 (2004).
14. T. Satoh, M. Miura, M. Nomura, J. Organomet. Chem. 653, 161 (2002).
15. R. Singh, S.P. Nolan, J. Organomet. Chem. 690, 5832 (2005).
16. A. Klapars, J.H. Waldman, K.R. Campos, M.S. Jensen, M. McLaughlin, J.Y.L. Chung, R.J. Cvetovich, C. Chen, J. Org. Chem. 70, 10186 (2005).
17. A.V. Mitin, A.N. Kashin, I.P. Beletskaya, Russ. J. Org. Chem. 40, 802 (2004).
18. D. Shabashov, O. Daugulis, Org. Lett. 7, 3657 (2005).
19. L. Golomba, E. Abele, R. Abele, P. Arsenyan, Pre OMCOS 13, Paris, 15-16 July, 2005, Recent advances in organometallic chemistry and applied catalysis, Abstracts, P 18.

Received on October, 27, 2011.